

## INSECT PHEROMONES AND THEIR ANALOGS.

LVII. SYNTHESIS OF THE RACEMIC ANALOG OF THE SEX  
PHEROMONE OF PINE SAWFLIES OF THE GENERA*Diprion* AND *Neodiprion*

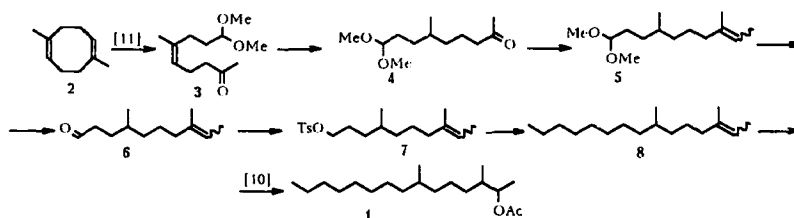
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A new route has been developed to the selective transformations of 1,1-dimethoxy-4-methyl-8-oxonon-4Z-ene — a product of the partial ozonolysis of 1,5-dimethylcycloocta-1Z,5Z-diene — leading to (±)-3,7-dimethylpentadec-2-yl acetate, the racemic analog of the sex pheromone of pine sawflies of the *Diprion* and *Neodiprion* genera.

Racemic (±)-3,7-dimethylpentadec-2-yl acetate (prionyl acetate) (1) exhibits the activity of the sex pheromone of four species of pine sawflies of the genera *Diprion* and *Neodiprion* [1, 2]. Several syntheses of the attractant (1) are known [2-9]. We have also reported the synthesis of compound (1) by selective transformations of 4-methyl-1,8-dioxonon-4Z-ene, including its selective reduction to a corresponding hydroxyketone and the protection of a keto group that is necessary in this case [10].

In the present communication we propose a shorter route to the synthesis of prionyl acetate (1) from 1,5-dimethylcycloocta-1Z,5Z-diene (2), the ozonolytic cleavage of which by a method described previously [11] leads to 1,1-dimethoxy-4-methyl-8-oxonon-4Z-ene (3). The use of the monoacetal (3) permitted a decrease in the number of stages and a doubling of the overall yield of the desired compound (1), which amounted to 20%, calculated on the initial isoprene dimer (2).



As the key stage in the proposed scheme of synthesis of the pheromone (1) we used the olefination of the saturated equivalent of the monoacetal (3) — 1,1-dimethoxy-4-methyl-8-oxononane (4), which is readily obtained by the catalytic hydrogenation of compound (3). Olefination of ketone (4) with ethylidetriphenylphosphorane gave 1,1-dimethoxy-4,8-dimethyldec-8-ene (5) in the form of a mixture (~4:1) of the Z- and E-isomers, as followed from a GLC analysis in a capillary column and from the ratio of the intensities of the singlets of the protons of the vinyl methyl group, CH<sub>3</sub>-8, resonating in the regions of  $\delta$  1.67 ppm (for the Z-isomer) and 1.54 ppm (for the E-isomer) [12]. Acid hydrolysis of the acetal (5) and reduction of the resulting 4,8-dimethyldec-8Z/E-enal (6) with a complex hydride (NaBH<sub>4</sub>) led to the alcohol corresponding to aldehyde (6), which was converted into the tosylate — 1-tosyloxy-4,8-dimethyldec-8Z/E-ene (7). The Li<sub>2</sub>CuCl<sub>4</sub>-catalyzed coupling of the latter with amylnmagnesium bromide took place smoothly, with the formation of 3,7-dimethylpentadec-2E/Z-ene (8) the hydroxylation of which under conditions found earlier [10] led to 3,7-dimethylpentadecan-2-ol, which was acetylated to the desired compound (1).

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## EXPERIMENTAL

IR spectra were taken on a UR-20 spectrometer in a thin layer. PMR spectra were obtained on a Tesla BS-567 spectrometer (working frequency 100 MHz) with  $\text{CDCl}_3$  as solvent and TMS as internal standard. GLC was conducted on a Chrom-5 instrument with the stationary phase SE-30 (5%) on Chromaton N-AW-DMCS (0.16-0.20 mm), flame-ionization detector, working temperature 50-300°C, carrier gas helium. In addition, compound (5) was analyzed on a Carlo Erba instrument with a glass capillary column (25 × 0.2 mm [sic]), working temperature 50-200°C. The elementary analysis of the compounds synthesized corresponded to the calculated values.

**1,1-Dimethoxy-4-methyl-8-oxononene (4).** A solution of 10.7 g (0.05 mole) of the monoacetate (3) in 600 ml of absolute ethanol-hexane (1:1) was hydrogenated (catalyst 5% Pd/C, ~25°C, normal pressure, 5 days) and, after filtration, the solution was passed through a 2.5-cm layer of  $\text{SiO}_2$  and evaporated. This gave 10.5 g (97%) of compound (4), bp 108-110°C (2 mm),  $n_D^{20}$  1.4383. The IR and PMR spectra were identical with those given earlier [13].

**1,1-Dimethoxy-4,8-dimethyldec-8Z/E-ene (5).** To a suspension of 8.5 g (0.023 mole) of ethylidetriphenylphosphorane in 70 ml of absolute THF being stirred at -30°C under argon was added 17.6 ml of a 1.3 N solution of BuLi in hexane, and the mixture was stirred at -30°C for 0.5 h and was treated with a solution of 3.2 g (0.015 mole) of the ketoacetal (4) in 10 ml of THF. The reaction mixture was stirred at -30°C for 1 h and at 25°C for 15 h and was then diluted with 300 ml of hexane and filtered. The filtrate was washed with saturated aqueous NaCl, dried with  $\text{MgSO}_4$ , and evaporated in vacuum, and the residue was chromatographed on 100 g of  $\text{Al}_2\text{O}_3$  (act. grade II). Elution with hexane-ether (3:1) gave 0.32 g of the initial substance (4) and 2.8 g (84%) of compound (5).

IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 860 (C=C-H), 1065 (C-O), 1380 ( $\text{CH}_3$ ), 1670 (C=C).

PMR spectrum ( $\delta$ , ppm): 0.8 (m, 3H,  $\text{CH}_3$ -4), 1.26 (m, 8H,  $\text{CH}_2$ ), 1.54 s, 1.57 d, 1.65 d, and 1.67 s, (6H,  $\text{CH}_3$ -C=C), 1.7 (m, 1H, H-4), 2.07 (m, 2H, H-7), 3.32 (s, 6H,  $\text{OCH}_3$ ), 4.33 (t, 1H, J = 5 Hz, H-1), 5.2 (m, 1H, H-9).

**4,8-Dimethyldec-8Z/E-enal (6).** A solution of 2.5 g (0.011 mole) of the acetal (5), 3 ml of  $\text{H}_2\text{O}$ , and 0.8 g of pyridinium tosylate in 130 ml of acetone was boiled for 2.5 h and was then evaporated in vacuum and the residue was dissolved in 500 ml of diethyl ether. The solution was washed successively with saturated aqueous solutions of  $\text{NH}_4\text{Cl}$ ,  $\text{Na}_2\text{CO}_3$ , and NaCl, dried with  $\text{MgSO}_4$ , and evaporated in vacuum. This gave 2.0 g (98%) of the aldehyde (6). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 860 (C=C-H), 1380 ( $\text{CH}_3$ ), 1670 (C=C), 1720 (C=O), 2720 (H-CO).

PMR spectrum ( $\delta$ , ppm): 0.84 (m, 3H,  $\text{CH}_3$ -4), 1.27 (m, 6H,  $\text{CH}_2$ ), 1.5-1.7 (m, 6H,  $\text{CH}_3$ -C=C), 1.7 (m, 1H, H-4), 2.0 (m, 2H, H-7), 2.3 (m, 2H, H-2), 5.2 (m, 1H, H-9), 9.5 (br.s, 1H, H-1).

**1-Tosyloxy-4,8-dimethyldec-8Z/E-ene (7).** At 20°C, with stirring, 0.65 g (0.017 mole) of  $\text{NaBH}_4$  was added to a solution of 1.8 g (0.01 mole) of the aldehyde (6) in 20 ml of absolute methanol and then the reaction mixture was stirred for 2 h and was left for 12 h, after which 15 ml of 5% AcOH was added, the methanol was distilled off, the residue was extracted with diethyl ether, and the extract was washed successively with water and saturated solutions of  $\text{Na}_2\text{CO}_3$  and NaCl, and was dried with  $\text{MgSO}_4$ . After evaporation, 1.7 g of a product ( $\nu_{\text{OH}}$  3300-3450 br.s) was obtained, and a solution of this in 2.8 ml of pyridine was treated at 0-5°C with 2.1 g (0.013 mole) of tosyl chloride. The reaction mixture was stirred for 5.5 h and was left at 8°C for 12 h. Then 40 ml of  $\text{Et}_2\text{O}$  was added to it and it was washed successively with 10% HCl and saturated solutions of  $\text{Na}_2\text{CO}_3$  and NaCl, and dried with  $\text{MgSO}_4$ . After evaporation in vacuum, 2.8 g (83%) of the tosylate (7) was obtained.

IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 1600 (C=C<sub>Ar</sub>).

PMR spectrum ( $\delta$ , ppm): 0.8 (m, 3H,  $\text{CH}_3$ -4), 1.2 (m, 8H,  $\text{CH}_2$ ), 1.5-1.7 (m, 6H,  $\text{CH}_3$ -C=C), 1.7 (m, 1H, H-4), 2.0 (m, 2H, H-7), 2.4 (s, 3H,  $\text{CH}_3$ -Ar), 3.96 (t, 2H, J = 6 Hz, H-1), 5.15 (m, 1H, H-9), 7.3 d and 7.75 d (4H, J = 8 Hz,  $\text{CH}_{\text{Ar}}$ ).

**3,7-Dimethylpentadec-2Z/E-ene (8).** At -78°C with stirring under Ar, 2.5 ml of a 0.1 M solution of  $\text{Li}_2\text{CuCl}_4$  in THF was added to a solution of 3.38 g (0.01 mole) of the tosylate (7) in 40 ml of absolute  $\text{Et}_2\text{O}$ , and then a solution of amylmagnesium bromide generated from 0.6 g (0.025 g-atom) of Mg and 2.8 g (0.018 mole) of  $n\text{C}_5\text{H}_{11}\text{Br}$  (with  $\text{CH}_3\text{I}$  as initiator) was added dropwise, and the resulting reaction mixture was stirred at -70°C for 2 h and was left at room temperature for 12 h, after which 20 ml of  $\text{Et}_2\text{O}$  was added and the mixture was stirred for 0.5 h and poured into a cold saturated solution of  $\text{NH}_4\text{Cl}$ . The mixture was shaken vigorously, and then the organic layer was separated off and was washed successively with saturated solutions of  $\text{NaHCO}_3$  and NaCl, dried over  $\text{MgSO}_4$ , and evaporated. The residue was chromatographed on 45 g of neutral  $\text{Al}_2\text{O}_3$  (hexane-ether (9:1)). This gave 0.25 g of *n*-decane and 1.95 g (82%) of the alkene (8),  $n_D^{20}$  1.4560. Its IR and PMR spectra were identical with those obtained in [10].

**3,7-Dimethylpentadec-2-yl acetate (1)**, obtained with a yield of 76% from the alkene (8) as described in [10], had IR and PMR spectra identical with those of the authentic compound.

## REFERENCES

1. D. Jewett, F. Matsumura, and H. C. Coppel, *J. Chem. Ecol.*, **4**, No. 3, 277 (1978).
2. R. Baker and P. M. Winton, *Tetrahedron Lett.*, **21**, No. 12, 1175 (1980).
3. K. Mori, S. Masuda, and M. Matsui, *Agric. Biol. Chem.*, **42**, No. 5, 1015 (1978).
4. P. Place, M.-L. Roumestant, and J. Yore, *J. Org. Chem.*, **43**, No. 5, 1001 (1978).
5. G. Magnusson, *Tetrahedron*, **34**, No. 9, 1385 (1978).
6. J. Kallmerten and M. Balestra, *J. Org. Chem.*, **51**, No. 14, 2855 (1985).
7. É. P. Serebryakov and G. D. Gamalevich, *Izv. Akad. Nauk SSSR, Ser. Khim.*, No. 1, 114 (1987).
8. V. N. Odínokov, G. Yu. Ishmuratov, R. Ya. Kharisov, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 573 (1989).
9. V. N. Odínokov, G. Yu. Ishmuratov, I. M. Ladenkova, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 818 (1990).
10. V. N. Odínokov, V. R. Akhmetova, G. Yu. Ishmuratov, L. P. Botsman, and G. A. Tolstikov, *Zh. Org. Khim.*, **22**, No. 5, 953 (1986).
11. V. N. Odínokov, V. R. Akhunova, R. S. Bakeeva, R. I. Galeeva, A. V. Semenovskii, A. M. Moiseenkov, and G. A. Tolstikov, *Zh. Org. Khim.*, **13**, No. 3, 532 (1977).
12. G. A. Tolstikov, V. N. Odínokov, V. R. Akhunova, R. I. Galeeva, U. M. Dzhemilev, A. M. Moiseenkov, and A. V. Semenovskii, *Izv. Akad. Nauk, SSSR, Ser. Khim.*, No. 4, 887 (1978).
13. V. N. Odínokov, O. S. Kukovinets, R. A. Zainullin, V. R. Sultanmuratova, and G. A. Tolstikov, *Zh. Org. Khim.*, **24**, No. 7, 1365 (1988).